Effect of the alkyl spacer length on the electrical performance of diketopyrrolopyrrole-thiophene vinylene thiophene polymer semiconductors†

Hojeong Yu,a Kwang Hun Park,c Inho Song,a Myeong-Jong Kim,c Yun-Hi Kimªd and Joon Hak Ohªa

For systematic investigation of the structure–property relationship, a series of diketopyrrolopyrrole-thiophene vinylene thiophene (DPP-TVT) polymers, ranging from 25-DPP-TVT to 32-DPP-TVT, with branched alkyl groups containing linear spacer groups from C2 to C9, has been synthesized. The electrical performance of these polymers is clearly dependent on the length of the spacer group and shows the odd–even alterations in chemical and electronic properties induced by the different alkyl chain spacers. Spacer groups with even numbers of carbon atoms exhibit higher charge-carrier mobilities than those with odd numbers of carbon atoms for the linear spacer groups from C2 to C7. Furthermore, the optimal side chain geometry in the DPP-TVT system for the most efficient charge transport contains the C6 spacer between the branching point and the backbone, showing the maximum hole mobility of 8.74 cm² V⁻¹ s⁻¹ (at VGS = −100 V). The results obtained herein demonstrate the intriguing odd–even effects induced by the length of the side chain alkyl spacers for DPP-TVT polymers, and provide insight into the side chain engineering for the most efficient charge transport in DPP-based polymer semiconductors.

Introduction

Optimal molecular design of polymer semiconductors via the molecular backbone and side chain engineering has led to remarkable advances in high-performance organic field-effect transistors (OFETs), surpassing the charge carrier mobility of amorphous silicon semiconductors.1–3 State-of-the-art, high-performance polymer semiconductors have been mainly obtained by molecular design for the conjugated backbone unit, mostly adopting alternating electron donor (D) and acceptor (A) building blocks that can enhance intermolecular interactions and intramolecular charge transfer. In recent years, among high-performance polymer backbones, the electron-accepting diketopyrrolopyrrole (DPP) moiety has attracted great interest in constructing high-performance donor–acceptor (D–A) type polymer semiconductors,4–8 showing charge carrier mobilities over 10 cm² V⁻¹ s⁻¹.9–11 The rational design of molecular backbones in polymer semiconductors is of great importance, as it intrinsically determines the electronic properties of organic semiconductors.12–19

Recently, however, it turned out that sophisticated side chain engineering can significantly enhance the charge carrier mobility of p-type,20–22 n-type,23,24 and ambipolar conjugated polymers.25 This is because the length,25–27 branching point position,12,20,28 and bulkiness29 of side chains can also critically affect the physicochemical properties of polymers (i.e., solubility, optical absorption capacity, and solid state packing) and consequently influence molecular packing and the π-planar distance closely related to the charge transport.1,30,31 In particular, the number of carbons in the alkyl side chain and/or spacers results in different molecular tilt angles on the deposited substrate. Therefore, depending on the odd or even characteristic of the number of carbons, the extension directions of the side chains in molecular packing may be determined, and the electrical properties of the resulting polymers may be systematically changeable. Such odd–even effects on molecular conformations and packing structures have been widely observed over organic thin films in chemical and/or physical aspects.1,32,33 Especially, odd–even...
alterations in chemical and electronic properties of organic thin films have originated from various complex interactions between organic molecules and surfaces (i.e., self-assembled monolayers) or intra/intermolecular interactions. This is because, depending on the odd or even number of structural units, molecular structural orientations follow the most energetically favorable arrangement. To date, most of odd–even studies have been limited to small molecular organic semiconductors since their packing orientations are simply predicted by theoretical packing analysis. Furthermore, the influence of alkyl chain spacers in the side chains has been mostly investigated using a small number of organic semiconductors containing relatively short alkyl spacers. Odd–even effects in the polymeric system have not been widely investigated so far, due to the possibility of free motion of polymer chains.

However, very recently, systematic studies on the effect of side chain branching positions have been reported in polymer semiconductors containing more than six polymers in series. In our previous study, we have observed the odd–even effect of the alkyl side chain spacers in a series of diketopyrrolopyrrole-vinylene selenophene (DPP-SVS) polymers on molecular packing and morphology, which subsequently affected the charge transport. In this regard, studies on the structure–property relationship of polymer semiconductors with a broad scope of the polymer series are highly desirable for the in-depth understanding of the intriguing odd–even effects as well as for providing a molecular design guideline for high-performance polymer semiconductors.

Herein, a new series of poly(diketopyrrolopyrrole-thiophene vinylene thiophene) (PDPP-TVT) polymers with branched alkyl groups containing linear spacer groups (C2 to C9), ranging from 25-DPP-TVT to 32-DPP-TVT, has been synthesized to systematically explore the possibility of the odd–even effect on the polymer semiconductor system. The D–A structures in DPP-TVT polymers allow efficient intramolecular charge transfer and facilitate intermolecular interactions. In addition, the TVT unit enhances the coplanarity of the conjugated π-backbones for effective molecular packing. The relationship between molecular structure and electrical properties in DPP-based polymer semiconductors has been investigated. DPP-TVT polymer thin films have been prepared by various solution-based coating methods, such as spin-coating, drop-casting, and solution-shearing techniques. Our results clearly revealed the existence of the odd–even effect on molecular packing, morphology, and electronic properties in DPP-TVT polymer thin films. Intriguingly, linear alkyl spacer groups with even numbers (C2, C4, and C6) showed shorter lattice spacing values than the alkyl spacer groups with odd numbers (C3, C5, and C7), leading to higher mobilities and well-connected nanoscale granular domains in thin film morphologies. 31-DPP-TVT and 32-DPP-TVT with the alkyl spacers of C8 and C9 did not show a clear odd–even effect, due to increased unpredictable intermolecular interactions between the long alkyl side chains. Among the series of DPP-TVT polymers, 29-DPP-TVT with the alkyl spacer of C6 exhibited the maximum hole mobility of 8.74 cm² V⁻¹ s⁻¹, due to the optimized π-planar packing and thin-film morphologies. These findings clearly demonstrate the existence of the odd–even effect of the alkyl spacer length on the electrical performance of DPP-based polymer semiconductors, and provide molecular design guidelines for side chains of high-performance polymer semiconductors.

Results and discussion

Synthetic strategies, synthesis, and characterization

A series of DPP-TVT polymers (25-DPP-TVT to 32-DPP-TVT) was synthesized in chlorobenzene solvent for 48 h by a Stille coupling reaction using Pd2(dba)3 and P(o-Tol)3, as shown in Scheme 1. The obtained polymers were purified by precipitation and successive Soxhlet extraction using methanol, hexane, and chloroform to remove residual monomers and oligomers. The structures of DPP-TVT polymers (25-DPP-TVT to 32-DPP-TVT) were characterized by ¹H-NMR analysis, and elemental analysis of the polymers was also performed (Fig. S1 and S2 of the ESI†). The polymers had good solubility in common organic solvents such as chloroform and chlorobenzene. The average molecular weights of the synthesized polymers were measured by gel permeation chromatography (GPC) using polystyrene standards in o-chlorobenzene. The number average molecular weights (Mn) and polydispersity indices (PDI) of polymers are summarized in Table 1. Thermogravimetric analysis (TGA) of 25-DPP-TVT to 32-DPP-TVT revealed their good thermal stability with 5% weight decomposition up to 335–428 °C (see Fig. S3 and Table S1 of the ESI†). In differential scanning calorimetry (DSC) analysis, a distinct phase transition was observed for 25-DPP-TVT and 26-DPP-TVT at 270.9 °C and 278.7 °C, respectively, while...
The UV-vis absorption spectra of the DPP-TVT polymer series (25-DPP-TVT to 32-DPP-TVT) in chloroform solution and thin film are displayed in Fig. 1, with the data being summarized in Table 1. The polymers exhibit dual optical absorption from 300 to 1000 nm with small high- and intense low-energy bands attributed to the localized π-π* and internal charge transfer (ICT) transitions, respectively. The absorption maxima of polymer films showed slightly wider and broader absorptions compared with those of solution (Fig. 1). The solution spectra showed slightly red shifted absorption by approximately 10–12 nm when going from 25-DPP-TVT to 32-DPP-TVT, which may result from the reduced steric hindrance by introducing a larger linear spacer group from C2 to C7 (from 25-DPP-TVT to 30-DPP-TVT) exhibited broadly fluctuating (00) reflection peaks were observed with higher-order diffraction peaks up to (400) reflections, while the as-cast DPP-TVT films exhibited relatively weak and broad diffraction peaks (Fig. S6 and Table S2 of the ESI†). Interestingly, a series of DPP-TVT polymer films containing the spacer group from C2 to C7 (from 25-DPP-TVT to 30-DPP-TVT) exhibited broadly fluctuating (100) diffraction peaks as shown in Fig. 2a. Such an odd-even effect was clearly observed in the average 2θ and d-spacing values of the DPP-TVT polymer films (Fig. 2b) and the corresponding out-of-plane XRD data are summarized in Table 2. These results showed that the even-numbered linear spacer groups (25-DPP-TVT with C2, 27-DPP-TVT with C4, and 29-DPP-TVT with C6) had shorter d-spacing values than those of the polymers with odd-numbered linear spacer groups (26-DPP-TVT with C3, 28-DPP-TVT with C5, and 30-DPP-TVT with C7).

**Table 1** Summary of the optical and electrochemical properties, and the number average molecular weight ($M_n$) and polydispersity of polymers determined by using GPC with polystyrene as the standard.

<table>
<thead>
<tr>
<th>Polymer</th>
<th>$\lambda_{0-0}^{\text{Sol}}$ [nm]</th>
<th>$\lambda_{0-1}^{\text{Sol}}$ [nm]</th>
<th>$\lambda_{0-0}^{\text{film}}$ [nm]</th>
<th>$\lambda_{0-1}^{\text{film}}$ [nm]</th>
<th>Band gap (optical) [eV]</th>
<th>LUMO (optical) [eV]</th>
<th>HOMO (electrochemical) [eV]</th>
<th>$M_n$ [kDa] (PDI)</th>
</tr>
</thead>
<tbody>
<tr>
<td>25-DPP-TVT</td>
<td>776 n</td>
<td>785 n</td>
<td>721 n</td>
<td>721 n</td>
<td>1.31</td>
<td>-3.97</td>
<td>-5.28</td>
<td>633 (1.27)</td>
</tr>
<tr>
<td>26-DPP-TVT</td>
<td>784 n</td>
<td>802 n</td>
<td>n</td>
<td>726 n</td>
<td>1.23</td>
<td>-4.03</td>
<td>-5.26</td>
<td>348 (3.28)</td>
</tr>
<tr>
<td>27-DPP-TVT</td>
<td>779 742 n</td>
<td>781 800 n</td>
<td>728 n</td>
<td>734 n</td>
<td>1.29</td>
<td>-3.98</td>
<td>-5.27</td>
<td>284 (2.25)</td>
</tr>
<tr>
<td>28-DPP-TVT</td>
<td>778 n</td>
<td>778 n</td>
<td>n</td>
<td>726 n</td>
<td>1.37</td>
<td>-3.90</td>
<td>-5.27</td>
<td>49 (5.90)</td>
</tr>
<tr>
<td>29-DPP-TVT</td>
<td>799 745 n</td>
<td>800 n</td>
<td>734 n</td>
<td>727 n</td>
<td>1.30</td>
<td>-3.97</td>
<td>-5.25</td>
<td>519 (2.67)</td>
</tr>
<tr>
<td>30-DPP-TVT</td>
<td>795 n</td>
<td>795 n</td>
<td>n</td>
<td>726 n</td>
<td>1.27</td>
<td>-4.00</td>
<td>-5.27</td>
<td>44 (1.26)</td>
</tr>
<tr>
<td>31-DPP-TVT</td>
<td>786 n</td>
<td>787 n</td>
<td>727 n</td>
<td>736 n</td>
<td>1.32</td>
<td>-3.94</td>
<td>-5.26</td>
<td>179 (3.22)</td>
</tr>
<tr>
<td>32-DPP-TVT</td>
<td>791 743 n</td>
<td>791 n</td>
<td>n</td>
<td>736 n</td>
<td>1.30</td>
<td>-3.97</td>
<td>-5.27</td>
<td>242 (2.39)</td>
</tr>
</tbody>
</table>

$^a$ UV-absorption for solution was measured by using a chloroform solution of $10^{-4}$ M. $^b$ UV-absorption for film was measured by spin coating using 0.2 wt% solution. $^c$ CV was measured in a solution containing Bu$_4$NClO$_4$ (0.1 M) in acetonitrile. $^d$ GPC was measured by polystyrene standard in chlorobenzene solvent. $^e$ Not observed.

No phase transition was observed from 50 to 300 °C for 27-DPP-TVT to 32-DPP-TVT (see Fig. S4 and Table S1 of the ESI†).

**Microstructural analysis of polymer films**

The molecular packing and crystallinity of the DPP-TVT polymer series (25-DPP-TVT to 32-DPP-TVT) in the film state were investigated by performing out-of-plane X-ray diffraction (XRD) analysis. In the annealed polymer films (200 °C for 10 min), well-defined strong (000) reflection peaks were observed with higher-order diffraction peaks up to (400) reflections, while the as-cast DPP-TVT films exhibited relatively weak and broad diffraction peaks (Fig. S6 and Table S2 of the ESI†). Interestingly, a series of DPP-TVT polymer films containing the spacer group from C2 to C7 (from 25-DPP-TVT to 30-DPP-TVT) exhibited broadly fluctuating (100) diffraction peaks as shown in Fig. 2a. Such an odd-even effect was clearly observed in the average 2θ and d-spacing values of the DPP-TVT polymer films (Fig. 2b) and the corresponding out-of-plane XRD data are summarized in Table 2. These results showed that the even-numbered linear spacer groups (25-DPP-TVT with C2, 27-DPP-TVT with C4, and 29-DPP-TVT with C6) had shorter d-spacing values than those of the polymers with odd-numbered linear spacer groups (26-DPP-TVT with C3, 28-DPP-TVT with C5, and 30-DPP-TVT with C7).

![Fig. 1](a) UV-vis-NIR absorption of the series of DPP-TVT polymers in (a) chloroform solution and (b) thin films.
This odd–even trend of molecular packing in polymer semiconductors suggests alternation in the molecular packing densities depending on the carbon number of spacer groups (not significantly affected by the donor group change in the DPP-based D–A system). DPP-TVT polymers have linear alkyl chains (C2–C9) as spacers and branched even-numbered n-alkane chains [C12H25 and C10H21] at the end group position. It is known that the odd/even number of alkyl spacers can determine the extension directions of chains.32,43 Whether the alkyl spacer has an odd/even number of CH2 groups, a definite odd–even difference in chain conformation or hydrogen bonding between adjacent molecules has been obtained and such a principle has been observed from a few organic small molecules containing the R–(CH2)m–R (m = 2n or 2n + 1) group.32 Our DPP-TVT polymers also clearly showed an odd–even difference in d-spacings.

The polymers containing even-numbered linear spacer groups (25-DPP-TVT with C2, 27-DPP-TVT with C4, and 29-DPP-TVT with C6) showed denser molecular packing with smaller d-spacing values than the polymers containing odd-numbered linear spacer groups (26-DPP-TVT with C3, 28-DPP-TVT with C5, and 30-DPP-TVT with C7). However, for the linear alkyl spacers larger than C7, the odd–even effect disappeared due to the unpredictable increased intermolecular interactions of the alkyl side chains. In molecular packing structural analysis of DPP-TVT thin films, it is found that there was no distinguishable difference in odd–even characteristics in between DPP-TVT and DPP-SVS polymer series. The results described herein suggest that an odd–even effect is dominantly affected by side chains, and not by the donor group change in the backbone, since the zigzag-like d-spacing variations are mainly originated from the difference in the molecular packing density in the out-of-plane direction.

To further investigate the molecular packing characteristics of DPP-TVT polymer films, the morphologies of the DPP-TVT polymer films annealed at 200 °C were investigated using tapping-mode atomic force microscopy (AFM). Fig. 3 shows the AFM height images (5 μm × 5 μm scan) of polymer films. The corresponding phase images are presented in Fig. S7 of the ESI.† Overall, the polymer films containing even-numbered spacers (25-DPP-TVT with C2, 27-DPP-TVT with C4, and 29-DPP-TVT...
with C6) were composed of nanoscale granular domains with better intergrain connectivity than those in the polymer films containing odd-numbered spacers (26-DPP-TVT with C3, 28-DPP-TVT with C5, and 30-DPP-TVT with C7). The root-mean-square (rms) roughness values were 0.73, 1.35, 2.07, and 2.11 nm for the 25-, 27-, 29-, and 31-DPP-TVT polymer films, respectively. In contrast, the polymer films with odd-numbered spacers exhibited 1.2 times higher average rms roughness values (1.17, 2.17, 1.21, and 2.71 nm for the 26-, 28-, 30-, and 32-DPP-TVT polymer films, respectively). Thin films based on 26-DPP-TVT, 28-DPP-TVT, and 30-DPP-TVT molecules showed comparatively unfavorable interconnected grains with larger holes within domain–domain interconnection which can act as charge-trapping sites. Furthermore, compared with the other polymer films, higher rms roughness values with rather unconnected morphology features were observed from 31-DPP-TVT and 32-DPP-TVT polymer films. This might imply that the long alkyl chains partially collapsed over the surface and were more aggregated than the other polymer films. The differences in the inter-domain connectivity and morphology roughness, as well as in the molecular packing structures, were indicative of the existence of the odd–even effect of the alkyl chain spacer in the DPP-TVT thin-film microstructures.

Fabrication of solution-processed FETs and I–V characterization

To investigate the charge transport property of a series of DPP-TVT polymer films, we fabricated bottom-gate top-contact field-effect transistors (FETs). The polymer solution in chlorobenzene (0.2 wt%) was coated on n-octadecyltrimethoxysilane (OTS)-modified SiO2/Si substrates using spin coating, drop casting, and solution shearing methods to investigate the changes in the mobility as a function of alkyl chain spacers in FETs. In the solution shearing method,25,44,45 a small volume of polymer semiconductor solution was placed on the OTS-treated Si substrate preheated at a mild temperature of 60 °C. Then, the solution was covered by a shearing substrate and dragged at a controlled speed while keeping the solution between two Si wafers. The experimental details are included in the Experimental Section. The electrical performance of FETs based on the solution-processed DPP-TVT polymer films is summarized in Table 3. The DPP-TVT-FETs exhibited typical p-channel characteristics and the maximum and the average mobilities were obtained from more than 15 FET devices. Fig. 4 shows the average field-effect mobilities with standard deviation values for the solution-processed FETs based on the DPP-TVT polymer films. Among the various coating methods, the solution-sheared thin films exhibited the highest mobilities due to the aligned and crystalline nature of the solution-sheared films. The corresponding transfer and output characteristics are shown in Fig. S8 of the ESI.† Note that the hole mobilities were calculated in the saturation regime from the transfer curves using a sweep range of ~10 V at the gate voltage due to the rather nonlinear behaviors over the whole sweep range, which might result from several factors, i.e., charge scattering effects induced at a high gate voltage, contact resistance, gate leakage, or bias stress. Relatively distinct fluctuations in the mobilities were observed within the range of 25-DPP-TVT and 30-DPP-TVT, clearly revealing the odd–even dependence. By utilizing various coating methods with repeated measurements, we could confirm that such enormous odd–even variations in both molecular packing and electrical transport are not simply obtained from experimental
fluctuations in error ranges. In addition to the observation of the odd–even effect in the hole transport, the optimized side chain-containing molecule was found to be 29-DPP-TVT with a C6 spacer unit. The solution-sheared 29-DPP-TVT films showed the highest average hole mobility of 8.29 ± 0.46 cm² V⁻¹ s⁻¹, with the maximum mobility of 8.74 cm² V⁻¹ s⁻¹ at VGS = -100 V, due to their shorter d-spacing value and well-interconnected nanoscale granular domains observed in the AFM morphology image.

To explore the practical applications of the optimized 29-DPP-TVT, photo-responses of 29-DPP-TVT FETs were investigated upon on-and-off switching of polychromatic light (λ = 450–650 nm, λmax = 640 nm, and Pmax = 36 mW cm⁻²) in the transistor off-state (VGS = 0 V and VDS = -100 V) (Fig. S9 of the ESL†). The photogenerated charge carriers produced noticeable photocurrents in 29-DPP-TVT FETs. Their highly reproducible photosensitivity could substantiate the potential of 29-DPP-TVT for use in photo-sensing applications. In addition, 30-DPP-TVT FETs exhibited much lower performance in contrast to 30-DPP-SVS FETs (the average hole mobility of 30-DPP-TVT was 1.09 ± 0.08 cm² V⁻¹ s⁻¹, while that of 30-DPP-SVS FETs was 9.78 ± 1.64 cm² V⁻¹ s⁻¹ in our previous report¹⁰). This might result from the much larger D-spacing value of 30-DPP-TVT thin film than that of 30-DPP-SVS, leading to unfavorable charge transport in OFET performance (the d-spacing of 30-DPP-TVT was 31.07 Å, while the d-spacing of 30-DPP-SVS was 28.0 Å). On the other hand, slight variations in the mobility were observed by increasing the side chain length (number) above 30-DPP-TVT. Such variation in the odd–even effect was consistent with the results from XRD and AFM analyses.

### Experimental

#### AFM characterization

An Agilent 5500 (Agilent, USA) scanning probe microscope (SPM) equipped with a Nanoscope V controller was used to obtain AFM images of DPP-TVT thin films. AFM images were recorded in high-resolution tapping mode under ambient conditions.

#### FET fabrication and testing

FET devices with bottom-gate top-contact configuration were prepared to characterize the electrical performance of a series of DPP-TVT polymers. A highly n-doped (100) Si wafer (<0.004 Ω cm) with a thermally grown SiO2 (300 nm, C = 10 nF cm⁻²) was utilized as the substrate and dielectric. The surface of SiO2 was modified with the n-octadecyltrimethoxysilane (OTS) self-assembled monolayer (SAM) as previously reported.⁴⁴,⁴⁵ 3 mM of OTS solution in trichloroethylene was spin-coated on the piranha-cleaned wafer at 3000 rpm for 30 s. Then, the wafer was exposed to ammonia vapor for ~12 h to facilitate the formation of OTS SAM, followed by sonication cleaning, sequential washing, and drying. The contact angle on the hydrophobic OTS-modified wafer with DI water droplets was >106°. The DPP-TVT polymers were dissolved in chlorobenzene (0.2 wt%) and the film was prepared on the substrate by various solution-processing methods; spin-coating, drop-casting, and solution-shearing.⁴⁴,⁴⁵ In solution-shearing, the substrate temperature and shearing speed were optimized for

### Table 3

<table>
<thead>
<tr>
<th>Conditions</th>
<th>p-Channel</th>
<th></th>
</tr>
</thead>
<tbody>
<tr>
<td>Polymer</td>
<td>Films</td>
<td>μ_{h,max}</td>
</tr>
<tr>
<td>25-DPP-TVT</td>
<td>Spin-coated</td>
<td>2.62</td>
</tr>
<tr>
<td></td>
<td>Drop-cast</td>
<td>3.59</td>
</tr>
<tr>
<td></td>
<td>Solution-sheared</td>
<td>4.10</td>
</tr>
<tr>
<td>26-DPP-TVT</td>
<td>Spin-coated</td>
<td>0.63</td>
</tr>
<tr>
<td></td>
<td>Drop-cast</td>
<td>0.71</td>
</tr>
<tr>
<td></td>
<td>Solution-sheared</td>
<td>0.80</td>
</tr>
<tr>
<td>27-DPP-TVT</td>
<td>Spin-coated</td>
<td>2.81</td>
</tr>
<tr>
<td></td>
<td>Drop-cast</td>
<td>4.35</td>
</tr>
<tr>
<td></td>
<td>Solution-sheared</td>
<td>4.69</td>
</tr>
<tr>
<td>28-DPP-TVT</td>
<td>Spin-coated</td>
<td>0.23</td>
</tr>
<tr>
<td></td>
<td>Drop-cast</td>
<td>0.26</td>
</tr>
<tr>
<td></td>
<td>Solution-sheared</td>
<td>1.20</td>
</tr>
<tr>
<td>29-DPP-TVT</td>
<td>Spin-coated</td>
<td>5.24</td>
</tr>
<tr>
<td></td>
<td>Drop-cast</td>
<td>5.76</td>
</tr>
<tr>
<td></td>
<td>Solution-sheared</td>
<td>8.74</td>
</tr>
<tr>
<td>30-DPP-TVT</td>
<td>Spin-coated</td>
<td>0.77</td>
</tr>
<tr>
<td></td>
<td>Drop-cast</td>
<td>0.90</td>
</tr>
<tr>
<td></td>
<td>Solution-sheared</td>
<td>1.23</td>
</tr>
<tr>
<td>31-DPP-TVT</td>
<td>Spin-coated</td>
<td>3.60</td>
</tr>
<tr>
<td></td>
<td>Drop-cast</td>
<td>4.37</td>
</tr>
<tr>
<td></td>
<td>Solution-sheared</td>
<td>6.69</td>
</tr>
<tr>
<td>32-DPP-TVT</td>
<td>Spin-coated</td>
<td>3.27</td>
</tr>
<tr>
<td></td>
<td>Drop-cast</td>
<td>4.31</td>
</tr>
<tr>
<td></td>
<td>Solution-sheared</td>
<td>4.85</td>
</tr>
</tbody>
</table>

* The p-channel characteristics of DPP-TVT FETs were measured at VDS = -100 V. * The maximum mobility over 15 FET devices with a channel length (L) of ~50 μm and a channel width (W) of ~1000 μm. * The average mobility over 15 FET devices with a channel length (L) of ~50 μm and a channel width (W) of ~1000 μm. * The standard deviation.

Fig. 4 The average field-effect mobilities obtained from the annealed DPP-TVT thin-film transistors. The distinct odd–even trend with the side chain length is observed in the range of 25-DPP-TVT to 32-DPP-TVT.
to 60 °C and 0.12 mm s⁻¹, respectively for thin film formation (thickness < 30 nm). Then, the polymer thin-film was annealed on a hot plate at 200 °C for 10 min under a N₂ atmosphere. Gold contacts (40 nm) were thermally evaporated onto the polymer film to form source and drain electrodes with a channel length (L) of ~50 μm and a channel width (W) of ~1000 μm using a shadow mask. The electrical performance of FETs was measured in a N₂-filled glove box using a Keithley 4200 semiconductor parametric analyzer. The field-effect mobility was calculated in the saturation regime using the following equation:

\[ I_D = \frac{W}{2L} \mu C_i (V_{GS} - V_T)^2 \]

where \( I_D \) is the drain-to-source current, \( W \) and \( L \) are the semiconductor channel width and length, respectively, \( \mu \) is the mobility, and \( V_{GS} \) and \( V_T \) are the gate voltage and threshold voltage, respectively.

Conclusion

A series of DPP-TVT polymers ranging from 25-DPP-TVT to 32-DPP-TVT, with branched alkyl groups containing linear spacer groups from C2 to C9, was synthesized to systematically investigate the structure–property relationship, especially to study the odd–even effect induced by the number of carbon atoms in the alkyl spacer of the DPP-TVT polymer side chains. The solid-state molecular packing density and charge transport capability were clearly altered, depending on whether the number was odd or even. Linear alkyl spacer groups with even numbers of carbon atoms (C2, C4, and C6) exhibited shorter lattice spacings than did spacer groups with odd numbers of carbon atoms (C3, C5, and C7), resulting in higher hole mobilities. The optimal alkyl spacer length was confirmed by a series of DPP-TVT polymers and 29-DPP-TVT containing the C6 linear spacer from the Global PhD Fellowship.

Acknowledgements

This work was supported by the Center for Advanced Soft Electronics under the Global Frontier Research Program (2013M3A6A5073175 and 2013M3A6A5073172) of the Ministry of Science, ICT and Future Planning, and by the National Research Foundation of Korea (2014R1A2A2A01007467 and 2012R1A2A2A06047047). H.Y. acknowledges financial support from the Global PhD Fellowship.

References


This journal is © The Royal Society of Chemistry 2015

J. Mater. Chem. C, 2015, 3, 11697–11704 | 11703


